Kinetic Determination of Serum Glucose by Use of the Hexokinase/Glucose-6-phosphate Dehydrogenase Method

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Summary: We have developed a kinetic fixed-time approach for the quantitative determination of serum glucose by use of the hexokinase/glucose-6-phosphate dehydrogenase method. To achieve a large measuring range, we have apparently increased the Michaelis constant of glucose-6-phosphate dehydrogenase through addition of the competitive inhibitor ATP. By this means, serum samples with glucose concentrations up to 55.5 mmol/l could be analyzed without pre-dilution. The method has been adapted to the ENI GEMSAEC analyzer and to the LKB 2086 Mark II analyzer. It yielded satisfactory results with regard to precision. A comparison of the kinetic procedure with the manual end-point method showed good agreement. No interferences from hemoglobin, bilirubin, or lipemia have been observed.

Kinetische Bestimmung von Glucose im Serum mit der Hexokinase/Glucose-6-phosphat-Dehydrogenase-Methode


Introduction

Due to its specificity and its insensitivity to interference, the hexokinase/glucose-6-phosphate dehydrogenase method has found worldwide acceptance as a reference method for the determination of glucose in body fluids (1). It is based on the following reaction sequence in which the second step is responsible for the high specificity of the assay (2).

\[
\begin{align*}
D-Glucose + ATP & \rightarrow D-Glucose-6-phosphate + ADP \\
(\text{Hexokinase} & \quad \text{EC 2.7.1.1}) \\
\text{Glucose-6-P dehydrogenase} & \quad \text{EC 1.1.1.49} \\
D-Glucose-6-P + NADP^+ & \rightarrow D-Glucono-6-lactone-6-P + NADPH + H^+ 
\end{align*}
\]

The assay is commonly carried out as an end-point procedure (1,2). To improve the convenience of the method for the automated clinical laboratory, we studied the feasibility of a kinetic fixed-time approach which allows reduction of the reaction time and omission of the technically elaborate determination of the initial absorbance (3).

Because of the low Michaelis constant of glucose-6-phosphate dehydrogenase with respect to glucose-6-phosphate (4), the kinetic glucose assay using this enzyme and hexokinase is not, at first sight, suited for the determination of high glucose concentrations (3). However, following the theory of Michaelis and Menten, this situation can be improved by the addition of a competitive inhibitor which apparently increases the \(K_m\).
value of the enzyme (3, 5). Therefore, we first looked for a suitable compound which competitively inhibits glucose-6-phosphate dehydrogenase with respect to glucose-6-phosphate. The kinetic glucose assay was then performed in the presence of this inhibitor, using an ENI GEMSAEC analyzer and an LKB 2086 Mark II analyzer.

Materials and Methods

Apparatus

The automated methods were performed using an ENI GEMSAEC centrifugal analyzer (Electro-Nucleonics, 368 Passaic Ave., Fairfield, N.J. 07006) and a LKB 2086 Mark II kinetic analyzer (LKB Clinicon AB, S-16126 Bromma, Sweden). For the LKB 2086 Mark II method, samples and reagents were dispensed with an LKB 2075 diluter. Manual assays were done with a Model 1101 M spectrum line photometer which was connected with a Model 6511 recorder (Eppendorf Gerätebau Netheler & Hinz GmbH, D-2000 Hamburg 63).

Reagents

Hexokinase test kit "Gluco-quant", cat. No. 245178; ATP cat. No. 128 040, glucose and bilirubin were from E. Merck, D-6100 Darmstidt, Germany. D-glucose and bilirubin were from E. Merck, D-6100 Darmstidt, Germany. According to the manufacturer's instructions reagent 1 of the test kit was reconstituted to 200 ml with distilled water giving a concentration of 70 mmol/1 phosphate buffer (pH 7.7), besides the test kit contained > 100 kU/1 hexokinase and > 180 kU/1 glucose-6-phosphate dehydrogenase from Leuconostoc mesenteroides, cat. No. 165 883, were all from Boehringer Mannheim, P.O. Box 51, D-6800 Mannheim 31, Germany.

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For the GEMSAEC method, a single reagent was prepared by adding 2 ml of reagent 2 and 1.80 g (2.97 mmol) of ATP to 200 ml of reagent 1 solution. For the LKB 2086 Mark II method, 3.15 g (5.20 mmol) of ATP was added to 200 ml of reagent 1 solution. The starting reagent was obtained by mixing 20 ml of this solution with 2.0 ml of reagent 2. For the inhibition studies, an assay mixture was prepared by adding 20 U of glucose-6-phosphate dehydrogenase and varying ATP concentrations to 200 ml of reagent 1 solution. The reagents were stable for at least one day at 25 °C.

Results and Discussion

At present, in commercially available test kits for the determination of glucose according to the hexokinase end-point procedure, the glucose-6-phosphate dehydrogenase from Leuconostoc mesenteroides is commonly used. This enzyme is known to be inhibited by pyridoxal-phosphate (6), ATP and acyl-CoA derivatives (7). Stimulated by the fact that ATP is already employed as low-priced and stable competitive inhibitor for the kinetic determination of serum triglycerides (8) we preferably looked into the suitability of that substance for the kinetic glucose assay.

To determine the type of inhibition of glucose-6-phosphate dehydrogenase by ATP, we first studied the kinetics of this enzymatic reaction in the presence of increasing amounts of ATP. The results of the experiments are shown in figure 1, using a Lineweaver-Burk plot (9). As indicated by the common point of intersection of the curves on the 1/v axis, ATP exerts a competitive type of inhibition with respect to glucose-6-phosphate.

This finding suggested to us that, for the kinetic glucose assay, the K_m-value of glucose-6-phosphate dehydrogenase could be increased by simply elevating the ATP concentration usually employed for the end-point method. An ATP concentration of about 17 mmol/l in the reaction mixture was calculated to cause the required shift of the Michaelis constant (3). Figure 2 demonstrates that under these conditions, after a short lag phase, the overall reaction of the glucose assay followed pseudo-first-order kinetics with respect to glucose concentration. This, however, is known to be the basic requirement for a kinetic substrate determination by use of the fixed-time procedure (3).
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ATP [mmol/l]

28.5

19.0

14.8

9.5

1.9

0

0.1

0.2

0.3

1 [v] [µmol/min-1]

10

5

1 [S] [mmol-1·l]

Fig. 3. Fixed-time absorbance change as a function of glucose concentration, measured with the GEMSAEC.

Fig. 1. Inhibition of glucose-6-phosphate dehydrogenase by ATP.

Fig. 2. Time course of reaction. A absorbance at time t, $A_{\infty}$ absorbance after completion of reaction. Samples: glucose standards $\text{---}\rightarrow$ 2.78 mmol/l, $\text{---}\rightarrow$ 5.55 mmol/l, $\text{---}\rightarrow$ 11.10 mmol/l, $\text{---}\rightarrow$ 22.20 mmol/l. The response curves were recorded at $\lambda$ 365 nm using the GEMSAEC reagent.

Measuring range and sensitivity

We assayed dilutions of a stock solution of high purity D-glucose giving concentrations between 2.78 and 55.5 mmol/l. Figure 3 shows the results obtained with the GEMSAEC method. A straight line relationship was found to exist between the fixed-time absorbance changes and the respective glucose concentrations. Furthermore, the figure demonstrates that the sensitivity of the assay was sufficiently high in the normal range of glucose. With the LKB 2086 Mark II method, the measuring range was found to extend up to 44.4 mmol/l glucose.

Precision

Table 1 lists the within-run precision of the present method. The data were obtained by simultaneous analysis of aliquots of calibration standard, normal and supranormal human sera. Day-to-day precision was studied for six days by repeated analysis of the control serum Precinorm U (Boehringer Mannheim). Satisfactory results were obtained with both methods; GEMSAEC: mean = 5.45 mmol/l, CV = 2.5%; LKB 2086 Mark II: Mean = 5.38 mmol/l, CV = 3.4%.

<table>
<thead>
<tr>
<th>Method</th>
<th>Sample</th>
<th>No. tests</th>
<th>Glucose (mmol/l)</th>
<th>$\bar{x}$ ± s</th>
<th>CV (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>GEMSAEC</td>
<td>Standard</td>
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<td>5.57</td>
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<tr>
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<td>0.67</td>
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<tr>
<td></td>
<td>Serum</td>
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<td>29.82</td>
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<td>0.70</td>
</tr>
<tr>
<td>LKB 2086</td>
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<td>Mark II</td>
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<tr>
<td></td>
<td>Serum</td>
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<td>23.41</td>
<td>0.32</td>
<td>1.37</td>
</tr>
</tbody>
</table>

Tab. 1. Within-run precision of present method.
Accuracy

The correlation of the automated methods with the manual hexokinase end-point method was tested on 59 native human sera.

The results obtained with the GEMSAEC method are illustrated in figure 4. The calculated regression parameters indicate the good accuracy of the method. Equally satisfactory results were obtained with the LKB 2086 Mark II method: \( y = 0.981 x + 0.578, r = 0.997 \).

Fig. 4. Comparison of present GEMSAEC method (y) with manual hexokinase method (x) for glucose in native human sera. \( n = 59, y = 0.987x + 0.26, r = 0.998 \).

<table>
<thead>
<tr>
<th>Bilirubin added (μmol/l)</th>
<th>Hemoglobin added (μmol/l)</th>
<th>Glucose found* (mmol/l)</th>
</tr>
</thead>
<tbody>
<tr>
<td>-</td>
<td>10</td>
<td>10.32</td>
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<tr>
<td>20</td>
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<td>15</td>
<td>10.36</td>
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<tr>
<td>-</td>
<td>60</td>
<td>10.52</td>
</tr>
<tr>
<td>-</td>
<td>150</td>
<td>10.48</td>
</tr>
</tbody>
</table>

* mean values of triplicate determination

Similar correlations between the kinetic methods and the endpoint method were observed using plasma samples (EDTA, heparin, citrate).

Interferences

As illustrated in table 2, colored blood components did not interfere with the kinetic method. The samples used in this experiment were taken from a human serum pool and different amounts of bilirubin or hemoglobin were added. Also no interference was observed from lipemia.

Conclusions

The present study shows that the reference method for the determination of glucose, the hexokinase/glucose-6-phosphate dehydrogenase method, can be carried out kinetically over a wide range of glucose concentration if appropriate assay conditions are used. The kinetic approach, however, is especially suited for automated instruments. It allows a short reaction time. Neither analysis of reagent or sample blanks nor determination of initial absorbances is needed.

References